

## 2-Bromo-3-(4-chlorophenyl)-1-(2,4-dichloro-5-fluorophenyl)prop-2-en-1-one

V. Dhanasekaran,<sup>a</sup> D. Gayathri,<sup>a</sup>  
D. Velmurugan,<sup>a\*</sup> K. Ravikumar<sup>b</sup>  
and M. S. Karthikeyan<sup>c</sup><sup>a</sup>Department of Crystallography and Biophysics,  
University of Madras, Guindy Campus, Chennai  
600 025, India, <sup>b</sup>Laboratory of X-ray  
Crystallography, Indian Institute of Chemical  
Technology, Hyderabad 500 007, India, and  
<sup>c</sup>Department of Chemistry, Mangalore Univer-  
sity, Mangalore 574 199, India

Correspondence e-mail: d\_velu@yahoo.com

## Key indicators

Single-crystal X-ray study

T = 293 K

Mean  $\sigma(\text{C}-\text{C}) = 0.004 \text{ \AA}$ 

R factor = 0.039

wR factor = 0.106

Data-to-parameter ratio = 18.8

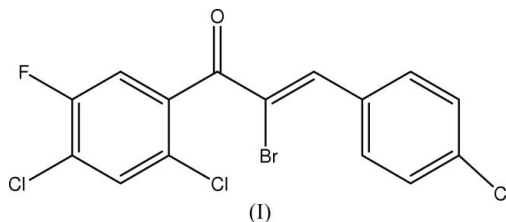
For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.The title compound,  $\text{C}_{15}\text{H}_7\text{BrCl}_3\text{FO}$ , crystallizes with two molecules in the asymmetric unit. The molecular conformation is stabilized by  $\text{C}-\text{H}\cdots\text{Br}$  intramolecular interactions.

Received 22 March 2007

Accepted 22 March 2007

## Comment

Chalcones are an important class of compounds that occur in edible plants. They have a wide variety of pharmaceutical activities including anticancer (Modzelewska *et al.*, 2006), anti-inflammatory (Won *et al.*, 2005) and antipyretic (De Leon *et al.*, 2003). Some chalcones show antiplasmodial (Go *et al.*, 2004) activity. If chalcones crystallize in a non-centrosymmetric space group they can be used as a non-linear optical device for second-harmonic generation. In view of the above, we have undertaken the X-ray crystal structure determination of the title compound (I).

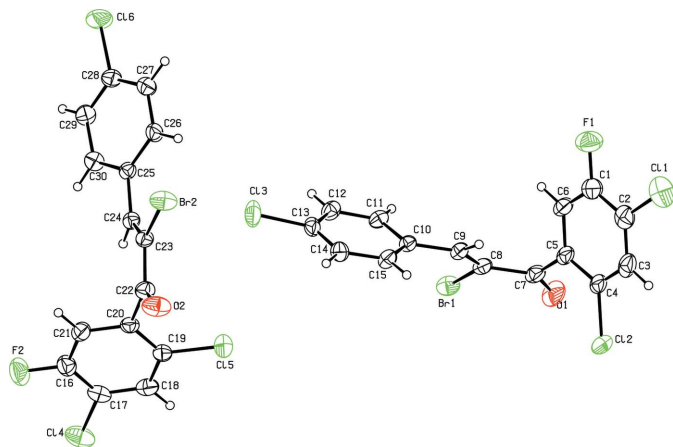


Bond lengths and bond angles of the title compound are comparable those in reported structures (Harrison *et al.*, 2006). The dihedral angle between the two benzene rings in one molecule is  $87.2(1)^\circ$  (C1–C6 and C10–C15) and in the other molecule is  $83.5(1)^\circ$  (C16–C21 and C25–C30). The O1–C7–C8–Br1 torsion angle is  $-4.1(4)^\circ$  in one molecule and  $0.2(4)^\circ$  (O2–C22–C23–Br2) in the other.

The molecular conformation is stabilized by  $\text{C}-\text{H}\cdots\text{Br}$  (Table 1) intramolecular interactions. An analysis of the crystal structure of (I) with *PLATON* (Spek, 2003) indicates no significant intermolecular hydrogen bonds.

## Experimental

3-(4-Chlorophenyl)-1-(2,4-dichloro-5-fluorophenyl)prop-2-en-1-one was prepared by a literature procedure (Shivarama Holla *et al.*, 2006). To a solution of propenone (1 mmol) in chloroform (25 ml), bromine (1 mmol) was added slowly with stirring and the reaction mixture was then stirred for 24 h. Excess chloroform was distilled off and the precipitated 2,3-dibromo-3-(4-chlorophenyl)-1-(2,4-dichloro-5-fluorophenyl)propan-1-one was filtered off and dried. A mixture of dibromopropanone (1 mmol) and triethylamine (1 mmol) in dry benzene (1 mmol) was stirred for 24 h. The excess solvent was removed under reduced pressure, giving the title compound which was recrystallized from chloroform by slow evaporation.



**Figure 1**  
The asymmetric unit of the title compound, showing 30% probability displacement ellipsoids.

#### Crystal data

$C_{15}H_7BrCl_3FO$	$V = 6064.6 (6) \text{ \AA}^3$
$M_r = 408.46$	$Z = 16$
Orthorhombic, $Pbca$	Mo $K\alpha$ radiation
$a = 14.3752 (9) \text{ \AA}$	$\mu = 3.24 \text{ mm}^{-1}$
$b = 11.3099 (7) \text{ \AA}$	$T = 293 (2) \text{ K}$
$c = 37.302 (2) \text{ \AA}$	$0.23 \times 0.22 \times 0.21 \text{ mm}$

#### Data collection

Bruker SMART APEX CCD area-detector diffractometer	7133 independent reflections
Absorption correction: none	4962 reflections with $I > 2\sigma(I)$
49754 measured reflections	$R_{\text{int}} = 0.042$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.039$	379 parameters
$wR(F^2) = 0.107$	H-atom parameters constrained
$S = 1.00$	$\Delta\rho_{\text{max}} = 0.86 \text{ e \AA}^{-3}$
7133 reflections	$\Delta\rho_{\text{min}} = -0.58 \text{ e \AA}^{-3}$

**Table 1**

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$C11-H11\cdots Br1$	0.93	2.89	3.297 (3)	108
$C26-H26\cdots Br2$	0.93	2.67	3.246 (3)	121

H atoms were positioned geometrically and were treated as riding, with  $C-H = 0.93 \text{ \AA}$  and  $U_{\text{iso}}(H) = 1.2U_{\text{eq}}(C)$ .

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINTE* (Bruker, 2001); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97* and *PARST* (Nardelli, 1995).

VD thanks DST for financial support. DV thanks the Department of Science and Technology (DST), India, for a major project.

#### References

- Bruker (2001). *SAINTE* (Version 6.28a) and *SMART* (Version 5.625). Bruker AXS Inc., Madison, Wisconsin, USA.
- De Leon, E. J., Alcaraz, M. J., Dominguez, J. N., Charris, J. & Terencio, M. C. (2003). *Inflamm. Res.* **52**, 246–257.
- Go, M. L., Wilairat, P., Rosenthal, P. J., Saliba, K. J. & Kirk, K. (2004). *Antimicrob. Agent. Chemother.* **48**, 3241–3245.
- Harrison, W. T. A., Yathirajan, H. S., Mithun, A., Narayana, B. & Sarojini, B. K. (2006). *Acta Cryst.* **E62**, o4508–o4509.
- Modzelewska, A., Catherine Petit, C., Achanta, G., Davidson, N. E., Huang, P. & Khan, S. R. (2006). *Bioorg. & Med. Chem.* pp. 3491–3495.
- Nardelli, M. (1995). *J. Appl. Cryst.* **28**, 659.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Shivarama Holla, B., Sooryanarayana Rao, B., Sarojini, B. K., Akberali, P. M. & Suchetha Kumari, N. (2006). *Eur. J. Med. Chem.* **41**, 657–663.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
- Won, S. J., Liu, C. T., Tsao, L. T., Weng, J. R., Ko, H. H., Wang, J. P. & Lin, C. N. (2005). *Eur. J. Med. Chem.* **40**, 103–112.